

# Standard Concentration

---

- $C_1V_1 = C_2V_2$

$C_1$  = Concentration of Solution 1

$V_1$  = Volume of Solution 1

$C_2$  = Concentration of Solution 2

$V_2$  = Volume of Solution 2



Be Right™

# Standard Concentration

---

- $C_1V_1 = C_2V_2$

$$C_1 = 27.62 \text{ mg/L}$$

$$V_1 = 0.2 \text{ mLs}$$

$$C_2 = \text{Unknown}$$

$$V_2 = 10.2 \text{ mLs (Sample + Standard)}$$



Be Right™

# Standard Concentration

---

- $C_1V_1 = C_2V_2$

$$C_1 = 27.62 \text{ mg/L}$$

$$V_1 = 0.2 \text{ mLs}$$

$$**C_2 = 0.54 \text{ mg/L}**$$

$$V_2 = 10.2 \text{ mLs (Sample + Standard)}$$



Be Right™